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562 Chapter 16 • Reaction Rates You can also choose to state the rate of the reaction as the rate at which CO is consumed, as shown below. average reaction rate = $\frac{[\text{CO}] \text{ at time } t_2 - [\text{CO}] \text{ at time } t_1}{t_2 - t_1} = \Delta[\text{CO}] / \Delta t$ Do you predict a positive or a negative value for this reaction rate?

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the unusual or exaggerated allergic reaction to an antigen that leads to severe respiratory and cardiac complications.

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a severe and rapid and sometimes fatal hypersensitivity reaction to a substance (especially a vaccine or penicillin or shellfish or insect venom) to which the organism has become sensitized by previous exposure

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Chapter 16 Chemical Kinetics □ Equilibrium vs. Thermodynamics vs. Kinetics □ Equilibrium constants help us predict how far a reaction will go (what it looks like) □ Thermodynamics allows us to predict the extent of a reaction, but it tells us nothing about the speed of the rxn or

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